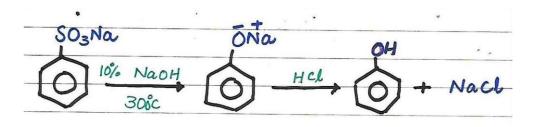
PREPARATION OF PHENOL

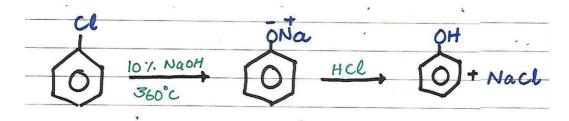
☐ REACTION OF SODIUM SALT OF BENZENE SULFONIC ACID WITH NaOH:

□ Sodium benzene sulfonate on fusion with strong alkali like NaOH at 300°C give sodium phenoxide which on treatment with HCl gives phenol



□ BASE HYDROLYSIS OF CHLOROBENZEN (DOW'S METHOD)

□ Chlorobenzene is hydrolysed by heating with 10% NaOH at 360°C under high pressure to form sodium phenoxide which on treating with HCl gives phenol



□ ACIDIC OXIDATION OF CUMENE

- ☐ It is recently developed commercial method for preparation of phenol. Cumene is oxidized by atmospheric oxygen is presence of metal catalyst into Cumene Hydroperoxide.
- ☐ The hydroperoxide is converted into phenol through acid catalyzed arrangement

 \square Aryl diazonium salts can be converted into phenols using $H_2O/H_2SO_4/$ heat